Protein-Modulated DNA Electron Transfer

Scott R. Rajski,[†] Sanjay Kumar,[§] Richard J. Roberts,[§] and Jacqueline K. Barton^{*,†}

The Division of Chemistry and Chemical Engineering California Institute of Technology Pasadena, California 91125 New England Biolabs, 32 Tozer Road Beverly, Massachusetts 01915

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Long-range oxidative damage to the 5'-guanine of 5'-GG-3' sequences in DNA readily occurs as a result of electron migration through the π -stacked DNA base pairs.^{1,2} Oxidative DNA damage has been demonstrated over distances of 200 Å,³ yet charge migration relies heavily upon the integrity of the DNA π -stack.⁴⁻⁶ The *HhaI* methyltransferase presents an opportunity to explore the effect of gaps within the π -stack on long-range electron transfer, since upon binding, the protein induces and stabilizes a π -gap using a novel DNA base-flipping mechanism.⁷

Long-range oxidation of 5'-GG-3' sites was first shown with a rhodium intercalator.^{8–12} The rhodium photochemistry bound to DNA yields base photooxidation upon irradiation at low energy (365 nm),^{8,13} whereas irradiation at high energy (313 nm)¹¹ leads to direct strand scission, marking the sites of intercalation. Other DNA-bound photooxidants also promote DNA damage at long range.^{4,6,14–17}

The methyltransferase *Hha*I–DNA complex is a wellcharacterized DNA–protein assembly in which the structure of DNA is significantly, but locally distorted.⁷ The enzyme recognizes 5'-G*CGC-3' and effects methylation by extruding the target cytosine (*C) completely out of the DNA duplex; the remainder of the B-form duplex is left intact. The DNA gap created by baseflipping is filled by insertion of the side chain of glutamine $237^{7,18,19}$ (Gln237).

Construction of a DNA assembly incorporating a M.*Hha*I binding site between two 5'-GG-3' doublets containing a tethered photooxidant would permit study of protein-induced π -stack distortions (Figure 1). Interruption of electron transfer would be

- (1) Holmlin, R. E.; Dandliker, P. J.; Barton, J. K. Angew. Chem., Int. Ed. Engl. 1997, 36, 2714.
- (2) Prat, F.; Houk, K. N.; Foote, C. S. J. Am. Chem. Soc. 1998, 120, 845.
 (3) Nunez, M. E.; Hall, D. B.; Barton, J. K. Chem. Biol. 1999, 6, 85.
- (4) Arkin, M. R.; Stemp, E. D. A.; Pulver, S. C.; Barton, J. K. *Chem. Biol.* **1997**, *4*, 389.
- (5) Hall, D. B.; Barton, J. K. J. Am. Chem. Soc. 1997, 119, 5045.
- (6) Gasper, S. M.; Schuster, G. B. J. Am. Chem. Soc. 1997, 119, 12762.
 (7) (a) Cheng, X.; Kumar, S.; Posfai, J.; Pflugrath, J. W.; Roberts, R. J.
- *Cell* **1993**, *74*, 299. (b) O'Gara, M.; Klimasauskas, S.; Roberts, R. J.; Cheng, X. J. Mol. Biol. **1996**, *261*, 634.
- (8) Hall, D. B.; Holmlin, R. E.; Barton, J. K. *Nature* 1996, *382*, 731.
 (9) Dandliker, P. J.; Holmlin, R. E.; Barton, J. K. *Science* 1997, *275*, 1464.
 (10) Johann, T. W.; Barton, J. K. *Philos. Trans. R. Soc. A* 1996, *354*, 299.
 (11) Sitlani, A.; Long. E. C.; Pyle, A. M.; Barton, J. K. *J. Am. Chem. Soc.*
- 1992, 114, 2303.
 (12) David, S. S.; Barton, J. K. J. Am. Chem. Soc. 1993, 115, 2984.
 (13) Turro, C.; Evenzahav, A.; Bossman, S. H.; Barton, J. K.; Turro, N.
- (15) Turto, C.; Evenzanav, A.; Bossman, S. H.; Barton, J. K.; Turto, N.
 Inorg. Chim. Acta **1996**, 243, 101.
 (14) Dandliker, P. J.; Nunez, M. E.; Barton, J. K. *Biochemistry* **1998**, 37,
- (14) Dandinker, P. J.; Nunez, M. E.; Barton, J. K. *Biochemistry* 1998, 57, 6491.
- (15) Hall, D. B.; Kelley, S. O.; Barton, J. K. *Biochemistry* 1998, *37*, 15933.
 (16) Schuster, G. B. Armitage, B.; Ly, D.; Koch, T.; Frydenlund, H.; Orum,
 H. *Proc. Natl. Acad. Sci. U.S.A.* 1997, *94*, 12320.
 (17) a) Nakatani, K.; Fujisawa, K.; Dohno, C.; Nakamura, T.; Saito, I.
- (17) a) Nakatani, K.; Fujisawa, K.; Dohno, C.; Nakamura, T.; Saito, I. *Tetrahedron Lett.* **1998** *39*, 5995. (b) Meggers, E.; Michel-Beyerle, M. E.; Giese, B. *J. Am. Chem. Soc.* **1998** *120*, 12950.
- (18) Mi, S.; Alonso, D.; Roberts, R. J. *Nucleic Acids Res.* 1995, 23, 620.
 (19) Garcia, R. A.; Bustamante, C. J.; Reich, N. O. *Proc. Natl. Acad. Sci.* U.S.A. 1996, 93, 7618.



Figure 1. DNA duplexes containing M.*Hha*I binding site 5'-GXGC-3' placed between two oxidizable 5'-GG-3' sites. Each duplex contains a Rh-tethered complement to the radiolabeled "target" strand. The complement was selectively functionalized⁹ at the 5' end with the Rh³⁺ intercalator and used as the racemic mixture of Δ and Λ diastereomers. Assemblies contained either matched **1A** or mismatched **1B** proteinbinding sites. For both sequence-coded (top) and space-filling diagrams (center), the 5'-GG-3' sites are shown in green with the M.*Hha* I site 5'-GCGC-3' in cyan. The photooxidant [Rh(phi)₂bpy']³⁺ (in red) is schematically intercalated into the DNA duplex at a site determined by photocleavage.

observed as differences in the extent of damage at the respective 5'-GG-3' sites. Two such substrates where the 5'-GG-3' doublets were incorporated on the strand complementary to the 5' Rh-tethered oligonucleotide were synthesized. These differed only in the M.*Hha*I DNA binding domain (DBD) **1A** and **1B**.²⁰

Upon photooxidation of these duplexes in the presence of protein, a significant decrease in distal 5'-GG-3' damage occurred in an enzyme-dependent fashion (Figure 2). Both **1A** and **1B** displayed diminutions of distal 5'-GG-3' damage at all enzyme concentrations; similarly, both DNAs showed significant protein binding in gel shift assays under parallel conditions.^{21–23} Interestingly, analysis of **1A** indicated an average diminution in distal G-oxidation of 42% and a corresponding formation of 25% reversibly gel-shifted DNA. Thus, in the case of **1A**, DNA-mediated charge migration revealed a higher degree of sensitivity to DNA-enzyme contacts than did the complementary gel-shift assay.

Additional alkali-labile damage at the 3'-G was found within the 5'-GUGC-3' of **1B** but not of **1A**; the distal/proximal damage ratio was comparable for the two DNAs, however. This damage was found to rely on enzyme, light, and Rh and was revealed only after piperidine digestion. A sensitivity to oxidation of the orphan G in the enzyme-bound mismatched site has been seen.²⁴ This damage event appears to be diagnostic of protein binding. Notably, we find that the extent to which both DNA damage events occur directly correlates with enzyme–DNA binding activity.

Gln237 insertion into the π -gap is a key feature of the DNA– protein base-flipped complex. We therefore examined also two

^{*} To whom correspondence should be addressed.

[†]California Institute of Technology.

[§] New England Biolabs.

⁽²⁰⁾ M.*Hha*I has high affinity for mismatches within its target sequence. See Klimasauskas, S.; Roberts, R. J. *Nucleic Acids Res.* **1995**, *23*, 1388. (21) The concentrations used were $\geq K_{\rm M}$ for M.*Hha*I binding to 5'-GCGC-

^{3&#}x27; and 5'-GUGC-3', assuming 100% enzymatic activity.²⁰ (22) The efficiency of photooxidative damage at the proximal 5'-GG-3'

site for both **1A** and **1B** was unchanged even with high M.*Hha*I concentrations (Figure 2). Also nonspecific competitor DNA enhances enzyme-dependent inhibition of distal G-oxidation.

⁽²³⁾ Trends in DNA damage observed at [DNA] of 100–200 nM were also unaffected by the addition of S-adenosyl-homocysteine; this analogue lacks the methyl group necessary for DNA methylation but is known to stabilize the DNA-M.*Hha*I complex.¹⁸

⁽²⁴⁾ Kim, J.; Sistare, M. F.; Carter, P. J.; Thorp, H. H. Coord. Chem. Rev. 1998,171, 341.



Figure 2. Photooxidative damage as a function of M.*Hha*I concentration and DBD sequence. The 5'-³²P-end-labeled DNAs as visualized by phosphorimagery following irradiation, piperidine digestion, and 20% denaturing polyacrylamide gel electrophoresis are shown. Lanes with Maxam–Gilbert sequencing reactions, each assembly (**1A** and **1B**) without irradiation (DC) and the duplex irradiated but without M.*Hha*I (BC) are as noted. Lanes 5–7 (**1A**) and 10–12 (**1B**) contained 100, 400, and 800 nM native M.*Hha*I, respectively. All samples contained 25 mM NH₄OAc (pH 9.0), 1 mM KHPO₄ (pH 7.4), 5 mM NaCl, 50 μ M EDTA, 5% glycerol, 200 nM SAH, 200 nM **1A** or **1B**, and a 40-fold molar excess (in base pairs) of poly dA–poly dT. Irradiations were performed for 1 h at 365 nm using a 1000-W Hg/Xe lamp with monochromator at 25 °C. The sites of rhodium intercalation appear as the band doubling near the native DNA band. Sites of proximal and distal 5'-GG-3' damage are shown.

mutants of M.*Hha*I in which Gln237 is replaced either with a purely aliphatic amino acid, isoleucine, or with the aromatic heterocyclic residue, tryptophan.²⁵ Insertion of Gln237 interrupts the aromatic manifold by which charge migration through the DNA can occur, while tryptophan insertion can be viewed as maintaining the π -stacked array. Isoleucine insertion would provide similar aliphatic—aromatic contacts to those of Gln237, but would lack the hydrogen-bonding network of native M.*Hha*I.

Figure 3 shows long-range oxidative damage of mismatch substrate **1B** with Ile237 and Trp237 M.*Hha*I mutants along with native M.*Hha*I. Here, optimally wild-type M.*Hha*I showed 45% inhibition of damage at long range. M.*Hha*I–Ile237 inhibited DNA-mediated electron transfer, although not to the same extent as native methylase. In parallel, much less damage at the orphaned 3'-G within the DNA-binding domain was observed with Ile237. Notably, both damage patterns for Ile237 indicate decreased complex formation compared to the wild type, which was confirmed by gel-shift assay.

More significantly, with Trp237, distal oxidative damage was retained, and hence the variant appeared to restore charge migration through the DNA duplex. The distal:proximal ratios for Trp237 closely paralleled the ratio obtained in the absence of



Figure 3. Photooxidative damage with bound mutants. (A) As in Figure 2 but showing only a closeup of DBD region using mismatched DNA 1B. (B) Plot of distal to proximal 5'-GG-3' damage ratio versus concentration of M.*Hha*I mutants.

enzyme. This retention of distal damage with Trp237 supports the notion that decreased distal damage by the wild type enzyme arises as a result of a protein-induced π -stacking disruption rather than through competitive hole-trapping by the protein. In fact, Trp radical formation would be thermodynamically feasible but appears kinetically not to yield significant trapped products.²⁶ That retention of long-range damage in the Trp237 mutant was not due to failure of the protein to bind is supported by an intense piperidine-labile G-reaction at the first 5'-G contained within the 5'-GUGC-3' site, as well as at the 3'-G (Figure 3). This enhanced 5'-G damage is unique to the tryptophan variant and relies upon the presence of Rh complex, light (365 nm) and piperidine digestion. Hence, this reaction provides another indication of intimate DNA-protein contact. This damage may be the result of a lowered oxidation potential for the 5'-G because of the inserted tryptophan to its 3'-side, as suggested by molecular modeling.² Importantly, this 5'-G damage as well as the retention of distal 5'-GG-3' oxidation both demonstrate that DNA-binding proteins can also activate long-range electron transfer through DNA.

These results indicate that long-range oxidative damage to DNA is highly sensitive to protein-induced DNA distortions. Here for the first time we observe that protein binding can modulate electron-transfer chemistry in DNA. Not only can protein–DNA interactions serve to inhibit electron transfer through the π -stack, but such interactions can also activate oxidative damage to DNA. These experiments demonstrate the potential utility of DNA electron transfer as a useful diagnostic and lend support to the intriguing idea that DNA-mediated charge transport may play a role in cellular events.

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⁽²⁵⁾ $K_{\rm M}$ values for the mutants are considered to be within an order of a magnitude to that of the wild-type based upon restriction assays. All mutants methylate DNA in vitro and in vivo.¹⁸

⁽²⁶⁾ $E^{\circ} = 1.2$ V for the Trp cation radical,²⁷ comparable to the guanine cation radical.⁴ Transient absorption studies of Lys-Trp-Lys bound to partially oxidized DNA duplexes indicate the formation of the Trp radical which decays on the ms time scale. Wagenknecht, H. A.; Stemp, E. D. A.; Barton, J. K., unpublished results.

⁽²⁷⁾ Meregnyi, G.; Lind, J.; Shen, X. J. Phys. Chem. 1988, 92, 134.